

Assignment: Subatomic Particles

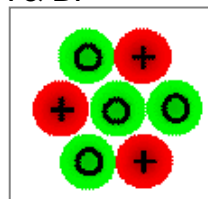
You will need a periodic table to help you answer these questions.

1. The nuclei of two atoms are shown in diagrams A & B.

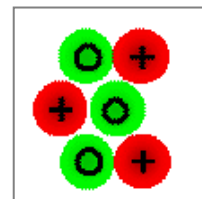
a. What elements are they?

A:

B:



A



B

b. What are their atomic weights?

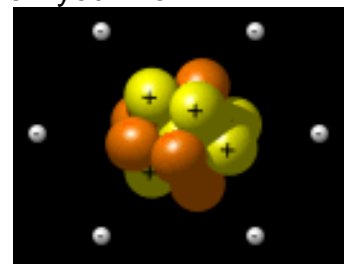
A:

B:

c. What is the word that describes what these atoms are?

2. Is the object in diagram C an atom, anion, or cation? Explain how you know.

a. What element is shown in diagram C?

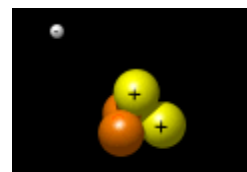


C

(there are 6 protons)

3. Is the object in diagram D an atom, anion, or cation? Explain why.

a. What element is shown in diagram D?



D

4. On the periodic table, find the element boron (B).

- a. How many protons does it have?
- b. How many electrons does it have?
- c. How many neutrons does it have?
- d. Draw a diagram of a boron atom below:

e. Draw a diagram of a boron anion with a charge of **-5**.

5. Draw two diagrams of aluminum below, each with the atomic mass given below.

Atomic Mass = 26

Atomic Mass = 28